

Article

Structural, Thermal and Magnetic Analysis of Two Fe-X-B (X = Nb, NiZr) Nanocrystalline Alloy

Kaouther Zaara, Jason Daza , Wael Ben Mbarek  and Joan-Josep Suñol * 

Department of Physics, Campus Montilivi s/n, University of Girona, 17003 Girona, Spain

* Correspondence: joanjosep.sunyol@udg.edu

Abstract: High-energy ball milling was used to produce two Fe-X-B (X = Nb, NiZr) nanocrystalline alloys. X-ray diffraction (XRD), differential scanning calorimetry (DSC), and vibrating sample magnetometry (VSM) were used to analyze the microstructure, thermal, and magnetic characteristics of the milled powders, the agglomerated particles (also generated during the milling process), and the compacted specimens of both alloys. The main crystallographic phase is always a bcc Fe-rich solid solution; whereas a minor Nb(B) phase is detected on powders and agglomerated particles in the Fe₈₀Nb₈B₁₂ alloy. The crystalline size of the Fe₈₀(NiZr)₈B₁₂ alloy is between 11 and 14 nm, whereas in the Fe₈₀Nb₈B₁₂ alloy, it ranges between 8 and 12 nm. Microstrain and dislocation density are higher in agglomerated samples for both alloys than in milled powders. Thermal analysis detects structural relaxation and crystal growth exothermic processes with high dispersion in the temperature intervals and in the calculated apparent activation energy of the main crystallization process. Regarding magnetic behavior, the coercivity values of all powdered-agglomerated specimens were around 800 A/m. The coercivity is higher in compacted sample, but controlled annealing favors enhanced soft behavior.

Keywords: mechanical alloying; nanocrystalline; structural analysis; magnetic analysis; thermal analysis



Citation: Zaara, K.; Daza, J.; Ben Mbarek, W.; Suñol, J.-J. Structural, Thermal and Magnetic Analysis of Two Fe-X-B (X = Nb, NiZr) Nanocrystalline Alloy. *Materials* **2023**, *16*, 155. <https://doi.org/10.3390/ma16010155>

Academic Editor: Tony Spassov

Received: 25 November 2022

Revised: 15 December 2022

Accepted: 21 December 2022

Published: 24 December 2022



Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

1. Introduction

Fe-based nanocrystalline alloys have received widespread research because of their higher magnetic properties and exceptional physical characteristics, such as low coercivity, high saturation magnetic flux density and effective permeability with respect to those exhibited by conventional microstructures [1,2]. Nanocrystalline Fe-M-B alloys, also known as Nanoperm-type alloys where M is an early transition metal, have attracted attention due to their remarkable soft magnetic characteristics. These alloys are employed as ultrasoft magnets in a variety of commercial applications, including telecommunications, microdevices, and power electronics [3]. The above alloys are often created as amorphous using rapid solidification processes; however, a nanocrystalline structure was detected following further annealing [4]. Another production method to directly manufacture nanocrystalline alloys with these compositions is mechanical alloying (MA) of elemental powders. Mechanical alloying has developed into a very powerful method to directly create metastable microstructures, such as nanocrystalline microstructures, supersaturated solid solutions, amorphous structures, etc. [5,6]. It has been discovered in the mechanically alloyed Fe_{94-x}Nb₆B_x alloys (x = 9, 14, 20), the Fe lattice parameter increases with higher B doping, whereas crystallite size is reduced [7]. In the mechanically alloyed Fe₆₂Nb₈B₃₀ powders, a combination of α-Fe, Nb(B), and highly disordered Fe (Nb, B) solid solution has been formed after 25 h, whereas the paramagnetic amorphous structure is attained after a longer milling time [8]. Mechanical alloying has been used to investigate the Fe₇₅Nb₁₀B₁₅ and Fe₈₅Nb₅B₁₀ systems. After milling, only Fe₇₅Nb₁₀B₁₅ alloy produces an amorphous

phase, whereas $\text{Fe}_{85}\text{Nb}_5\text{B}_{10}$ alloy produces a bcc supersaturated solid solution [9]. Al-leg et al. analyzed the magnetic and structural characteristics of mechanically alloyed $\text{Fe}_{75}\text{Si}_{15}\text{B}_{10}$ powders. They observed that dispersing B and Si into the Fe lattice results in the creation of bcc-Fe(Si,B) solid solution, with coercivity and magnetization values of 55 Oe and 6.7 emu/g, respectively, after 150 h of milling [10]. Chabi et al. investigated the influence of boron content on the microstructural, structural, and magnetic characteristics of $\text{Fe}_{92-x}\text{Nb}_8\text{B}_x$ ($x = 5, 10, 15,$ and 20) alloys and discovered FeB boride is seen for higher boron amounts ($x = 15$ and 20). According to their results, when the B concentration increases, the amorphous relative proportion and coercivity increase, while saturation magnetization decreases [11]. In another study by Mnasri et al., $\text{Fe}_{71}\text{Nb}_{23}\text{B}_6$ nanocrystalline alloy is produced by milling. They reported the main phase in both alloys is always bcc Fe-rich solid solution. Nonetheless, following milling, a small Nb(B) phase is found, and a minor amorphous phase is formed after 200 h of milling [12]. Likewise, Ipus et al. investigated the magnetic characteristics of FeNbB milled alloys with crystalline and amorphous boron. Their findings revealed amorphous boron was more effectively integrated into the matrix than crystalline boron, especially for short milling durations [13]. It is evident the preparation conditions affect the final product and the mechanically alloyed powders' properties. The mechanical alloying technique is characterized by the repeated welding and fracturing of powder particles in a high-energy ball mill, which frequently results in excessive cold welding and ductile particle aggregation [14]. Even though agglomeration is a significant issue in the preparation process when the particulate size is reduced to the micron level, Hong and Kao [15] successfully incorporated very fine SiC particulates with a mean particle size of 0.3 μm into an Al matrix using the mechanical alloying (MA) process, which was used to improve the agglomeration of the reinforcement particulates as well as to reduce their crystalline size. This method is basically a ball-milling process in which the powder particles are exposed to high-energy impact [16]. Since the presence of agglomerated particles was detected throughout the mechanical process in our study, a detailed evaluation of the agglomerated particles is required. As a result, the current work aims to investigate the morphology, structural, microstructural, thermal, and magnetic characteristics of the $\text{Fe}_{80}\text{-X}_8\text{-B}_{12}$ ($X = \text{Nb}, \text{NiZr}$) (at. %) alloy powders and the agglomerated particles formed by mechanical alloying for 40 and 80 h of milling. Nanocrystalline powders (80 h of milling) were compacted. Microstructure, thermal stability, and magnetic response were investigated before and after compaction. Finally, annealing was performed to improve the soft magnetic properties of the compacted specimens.

2. Materials and Methods

The mechanical alloying process was conducted in a high-energy planetary ball mill (Fritsch Pulverisette P7). The milling started with pure element and compound powders, Fe with a purity of 99.7% and a particle size of less than 8 μm ; Nb with a purity of 99.85% and a particle size of less than 74 μm ; B with a purity of 99.6% and a particle size of less than 50 μm ; and prealloyed $\text{Ni}_{70}\text{Zr}_{30}$ powders (purity of 99.9% and particle size greater than 150 μm). $\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$ and $\text{Fe}_{80}\text{Nb}_8\text{B}_{12}$ were the nominal compositions produced and evaluated. Milling was conducted during 40 and 80 h under an argon atmosphere with a ball to powder mass ratio of 5:1. To analyze the agglomerated particles formed after 40 and 80 h for each alloy, samples were separated and crushed into powders using a mortar and pestle. Likewise, the final powders (80 h of milling) were consolidated to obtain bulk specimens. In addition, taking into account thermal analysis data, annealing at selected temperatures was performed. Six samples of each composition were prepared (two before annealing, two after annealing at 300 °C, and two after annealing at 600 °C). Measurements were performed in open magnetic circuit by applying OIEC 60404-7 method B (to measure test specimens almost geometry independent).

Scanning electron microscopy (SEM) with a Zeiss DSM 960A instrument was used to examine the morphology of the MA powders and agglomerated particles (Zeiss, Oberkochen, Germany). The compositions of the specimens were checked by inductive coupled plasma

(ICP) in a Liberty-RL ICP Varian device. The powders, agglomerated particles, and compacted specimens were microstructurally characterized by x-ray diffraction (XRD) in a Bruker APEX D8 Advance diffractometer (Bruker, Billerica, MA, USA) using Cu-K radiation at a wavelength of $\lambda = 1.5406 \text{ \AA}$. The Rietveld refinement approach was used to refine the diffraction patterns by employing MAUD software (version 2.8). Thermal characterization was carried out using a Mettler-Toledo DSC30 device utilizing differential scanning calorimetry (DSC) under argon atmosphere in a temperature range from $100 \text{ }^\circ\text{C}$ to $600 \text{ }^\circ\text{C}$ at heating rates of 5, 10, 20, 30, and $40 \text{ }^\circ\text{C min}^{-1}$. All samples were magnetically characterized using magnetometry at room temperature with magnetic hysteresis loops in a SQUID MPMS-XL apparatus (Quantum Design; San Diego, CA, USA). The specimens analyzed and its labels are given in the Table 1.

Table 1. Samples analyzed in the research with the corresponding label. MA signifies mechanical alloying.

Samples	A = $\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$	B = $\text{Fe}_{80}\text{Nb}_8\text{B}_{12}$
MA for 40 h (powder)	AP-40	BP-40
MA for 40 h (agglomerate)	AA-40	BA-40
MA for 80 h (powder)	AP-80	BP-80
MA for 80 h (agglomerate)	AA-80	BA-80
MA for 80 h (compacted)	AC-80	BC-80

3. Results and Discussion

3.1. Morphology Analysis

The morphologies (SEM micrographs) for samples AP-40, AA-40, AC-80, BP-40, BA-40, and BC-80, are presented in Figure 1. A noticeable difference in the uniformity of the particle size (micrometric range) for powders and agglomerated particles of both alloys can be observed. The comparison was possible in sample A ($\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$) because the agglomerate particles deagglomerated easily, allowing the particle size distribution to be evaluated. The formation of agglomerates is caused by particle deformation and welding during milling.

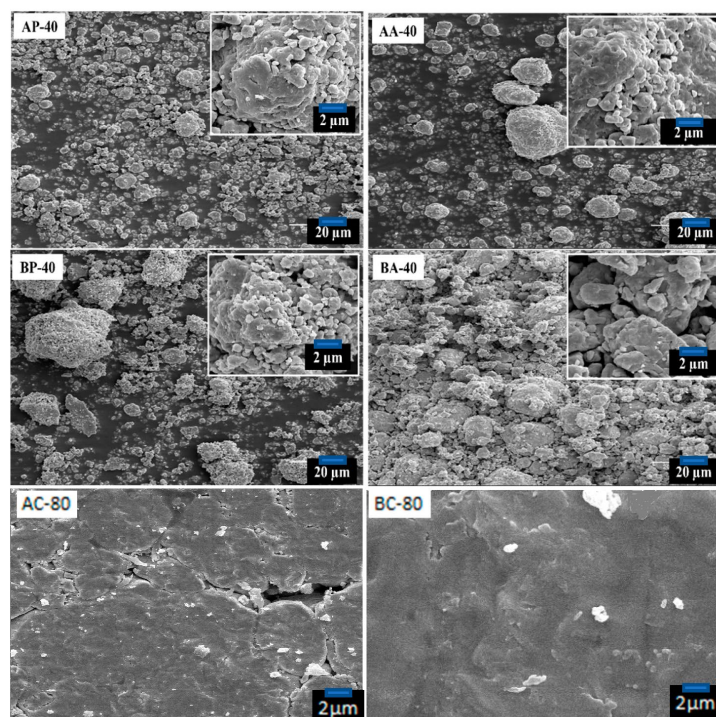


Figure 1. SEM images corresponding to samples AP-40, AA-40, BP-40, BA-40, AC-80, and BC-80.

The particle size distributions milling are given in Figure 2. First, for the AP-40, the majority of the particles are found below 3 μm with diameters of up to 10 μm . On the other hand, as predicted, there is a significant increase in particle diameter between the powder and agglomerate forms. For AA-40, the majority of the particles are found below 10 μm with diameters of up to 34 μm . Second, for B ($\text{Fe}_{80}\text{Nb}_8\text{B}_{12}$) powders milled for 40h (BP-40), the vast majority of particles are less than 4 μm . However, due to the sample's tendency for agglomeration, we find diameters of up to 46 μm in this case. For the same alloy in the agglomerated phase (BA-40), as expected and clearly visible on the SEM image, the majority of the particles are found below 15 μm with diameters of up to 55 μm . The agglomeration tendency is explained by the fact the Nb in this alloy has a low density, high ductility, and good formability in comparison to other elements [17]. Regarding the detected asymmetry of the distributions, the kurtosis values are 2.716, 1.501, 4.314, and 1.247 for alloys AP-40, AA-40, BP-40, and BA-40, respectively.

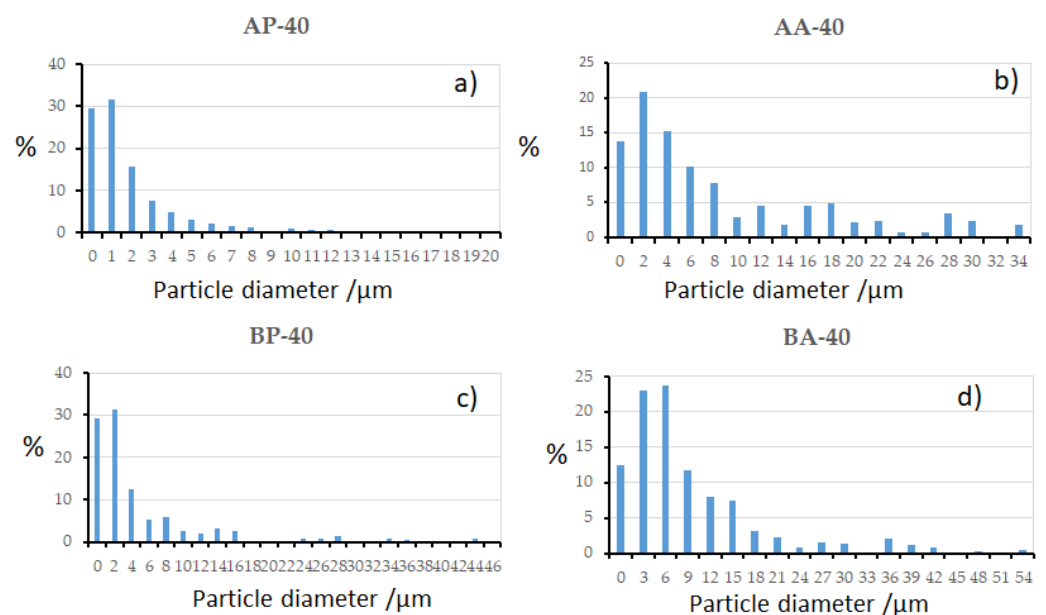


Figure 2. Particle size distribution of: (a) AP-40, (b) AA-40, (c) BP-40, and (d) BA-40.

Regarding the composition checked by ICP, the results show the composition ratios of Fe, Ni, Zr, Nb, and B are similar in powders and agglomerates to the nominal composition (and also in bulk specimens).

Lastly, as seen in the SEM micrographs of the compacted samples, the BC-80 exhibited a higher tendency to compact than the AC-80, which showed visible holes and pores.

3.2. XRD Analysis

Likewise, in terms of checking and analyzing, all samples for both alloys were subjected to XRD at room temperature. Figures 3 and 4 show the diffraction patterns of the obtained powders, agglomerated particles, and compacted specimens of A and B. The Rietveld refinement method was applied to samples with the scientific MAUD software (version 2.8, L. Lutterotti, Trento, Italy).

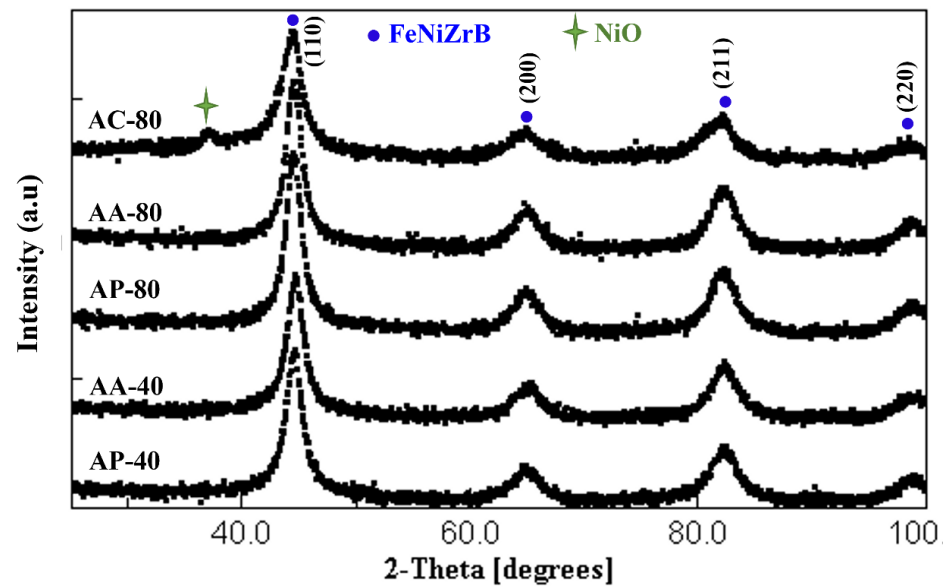


Figure 3. XRD diffractograms for samples of A alloy ($\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$).

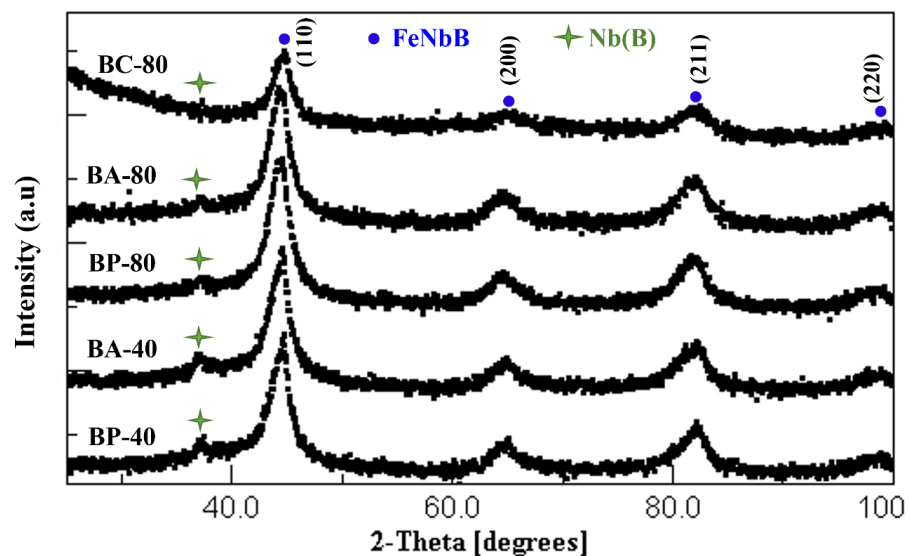


Figure 4. XRD diffractograms for samples of B alloy ($\text{Fe}_{80}\text{Nb}_8\text{B}_{12}$).

The Rietveld analysis results show the main crystallographic structure of all the samples examined corresponds to a body-centered cubic (BCC) nanocrystalline structure. The highest intensity peak was detected at an angle 2θ of about 44° . This peak is related to the α -Fe BCC structure (Ref. ICDD code: 01-087-0722). Only the α -Fe rich solid solution phase was found in alloy A milled after 40 and 80 h.

Milling often facilitates the appearance of Fe-B compounds in alloys with Fe and B; however, in Nb-B milling, the Nb(B)-rich phase identified in this study was also reported in the literature [12–18]. The reaction between B and Nb is attributed to the negative mixing enthalpy which is about -39 kJ/mol. Tables 2 and 3 display the cell parameter, crystal size, microstrain index (ϵ), and dislocation densities (ρ) for the A ($\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$) and B ($\text{Fe}_{80}\text{Nb}_8\text{B}_{12}$) sample alloys (powder, agglomerated, and compacted), respectively. Relevant refinement parameters are also included. These are the weighted residual (Rwp), expected residual (Rexp), and goodness of fit parameters (GoF).

Table 2. Crystallographic parameters and Rietveld refinement results obtained by Rietveld refinement of the A ($\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$) alloy.

Samples	a (Å)	Crystallite Size (nm)	ϵ (%)	ρ (10^{16} m^{-2})	Rwp (%)	Rexp (%)	GoF
AP-40	2.869(4)	11.98	0.33 (6)	0.69 (6)	16.50	15.257	1.081
AA-40	2.869(3)	11.19	0.53 (1)	0.79 (8)	17.456	16.27	1.072
AP-80	2.870(1)	14.47	0.55 (1)	0.47 (7)	14.920	13.552	1.100
AA-80	2.869(9)	14.05	0.61 (3)	0.50 (6)	16.153	14.396	1.121
AC-80	2.869(1)	26.11	0.30 (1)	0.14 (6)	17.258	13.585	1.270

Table 3. Crystallographic parameters and Rietveld refinement results obtained by Rietveld refinement of the B ($\text{Fe}_{80}\text{Nb}_8\text{B}_{12}$) alloy.

Samples	a (Å)	Crystallite Size (nm)	ϵ (%)	ρ (10^{16} m^{-2})	Rwp (%)	Rexp (%)	GoF
BP-40	2.876(1)	8.21	0.34 (3)	1.48 (3)	18.657	14.924	1.250
BA-40	2.876(9)	12.15	0.46 (8)	0.67 (7)	20.490	14.26	1.43
BP-80	2.876(4)	12.02	0.48 (7)	0.69 (2)	16.36	13.17	1.241
BA-80	2.880(9)	11.78	0.51 (1)	0.72 (1)	16.09	13.65	1.178
BC-80	2.879(7)	15.45	0.44 (6)	0.41 (8)	14.89	13.49	1.06

After 40 h of milling, the powder and agglomerated particles of the A ($\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$) alloy have nearly the same lattice parameter and crystallite size of 2.869 Å and 11 nm, respectively. The agglomerated particles AA-40, on the other hand, have a higher microstrain (ϵ) of about 0.53% and a higher value of dislocation densities (ρ) of about $0.79 \times 10^{16} \text{ m}^{-2}$. While the analyzed AP-40 has about 0.33% microstrain (ϵ) and a $0.69 \times 10^{16} \text{ m}^{-2}$ dislocation density (ρ). After 80 h of milling, the lattice parameters of the powder milled and agglomerated particles (AP-80 and AA-80) increased to approximately 2.870 Å and 2.869 Å, respectively. However, an increase in crystallite size was detected to be approximately 14 nm for both samples. Because no significant change in the temperature of the vials after ball milling was found, the increase in particle size after 80 h indicates the powders continue to cold-weld together during ball milling [19,20]. The measured microstrains (ϵ) were 0.55% and 0.61%, respectively, and their dislocation densities (ρ) were $0.47 \times 10^{16} \text{ m}^{-2}$ and $0.50 \times 10^{16} \text{ m}^{-2}$, respectively. According to the microstructure analysis of the A ($\text{Fe}_{80}(\text{NiZr})_8\text{B}_{12}$) alloy, the agglomerated particles have a higher microstrain and dislocation density than the powders milled after the same milling time, despite having nearly the same values of lattice parameter and crystallite size. The compact specimen of the powders milled after 80 h shows the lowest lattice parameter of nearly 2.869 Å and the highest crystallite size of about 26 nm. The dislocation density (ρ) and microstrain (ϵ) were reduced to 0.30% and $0.14 \times 10^{16} \text{ m}^{-2}$, respectively.

For the samples of the B ($\text{Fe}_{80}\text{Nb}_8\text{B}_{12}$) alloy milled for 40 h, the microstructure results indicate the agglomerated particles have a higher lattice parameter ~ 2.877 Å and a larger crystallite size (12.15 nm) than the powders milled (2.877 Å, 8 nm, respectively). This behavior is consistent with SEM images, which showed the agglomerated particles obtained after 40 h seemed to have large diameters. Consequently, despite having nearly the same crystallite size, the agglomerated particles have a higher lattice parameter (2.880 Å) than the milled powders (2.876 Å) after 80 h of milling. As with the A alloy, the agglomerated samples obtained after 40 and 80 h of the B alloy show greater microstrain and dislocation density than the milled powders. After 80 h, the compact specimen of the powders milled has a lattice parameter of nearly 2.879 Å and a crystallite size of about 15 nm. The microstrain (ϵ) and dislocation density (ρ) were reduced to 0.44% and $0.41 \times 10^{16} \text{ m}^{-2}$, respectively. For applications, the powdered mechanically alloyed samples should be

compacted to obtain bulk pieces. To preserve the functional properties, it is essential the compaction process does not affect the microstructure. The XRD analysis determined the nanocrystalline structure was retained. Nevertheless, a NiO minor phase $\sim 5.51\%$ was detected on the specimen compacted after milling the A alloy for 80 h (AC-80). Similarly, a minor phase associated to Nb(B)-rich solid solution is found in all B samples.

3.3. Thermal Analysis

To check the thermal stability of the nanostructured alloys, DSC analysis was performed, as shown in Figures 5 and 6. The general tendencies of the DSC traces for the powders and agglomerated particles are similar for each milling time. On the one hand, exothermic processes at low temperatures (below $400\text{ }^{\circ}\text{C}$) in alloy A (AP-40, AA-40, AP-80, and AA-80) were associated to relaxation phenomena generated by a reduction in free volume when mechanically induced tensions were relaxed at micro and nanoscale. Moreover, thermal treatment promotes atomic diffusion, which reduces local inhomogeneity and crystallographic defects. On the other hand, the same relaxation behavior was exhibited in the alloy B (BP-40, BA-40, BP-80, and BA-80) at low temperatures (between 300 and $400\text{ }^{\circ}\text{C}$). Higher temperature exothermic processes seen in both alloys were attributed to crystallization (crystal growth and/or nucleation). This complicated behavior is characteristic of mechanically alloyed nanocrystalline alloys [21] and is favored by non-homogeneity when Ni or Co are added to Fe-rich alloys. The peak temperature of the powders AP milled after 40 h is lower ($T_p \sim 484\text{ }^{\circ}\text{C}$) than the temperature of the agglomerated particles AA produced after the same milling time ($T_p \sim 495\text{ }^{\circ}\text{C}$). After 80 h, the peak temperatures of the powders and agglomerated particles were nearly identical at $483\text{ }^{\circ}\text{C}$ and $482\text{ }^{\circ}\text{C}$, respectively. For the alloy B, the powders milled and the agglomerated particles obtained after 40 and 80 h did not show a noticeable difference. For all samples of the B alloy, a wide exothermic process beginning at $400\text{ }^{\circ}\text{C}$ might be caused by early surface crystallization (particle surface) combined with stress [22].

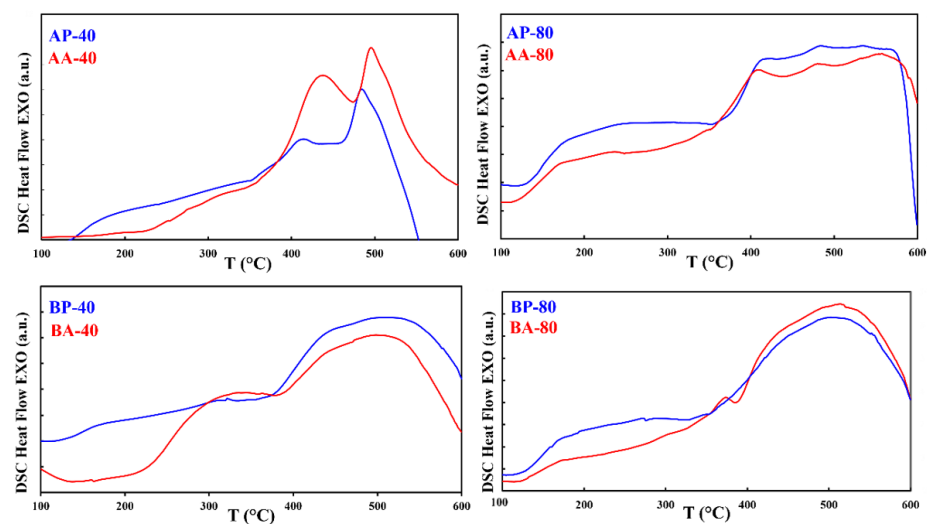


Figure 5. DSC curves collected in pairs for samples (AP-40 and AA-40), (AP-80 and AA-80), (BP-40 and BA-40), and (BP-80 and BA-80) to observe the behavior of powders and agglomerated particles after the same milling duration.

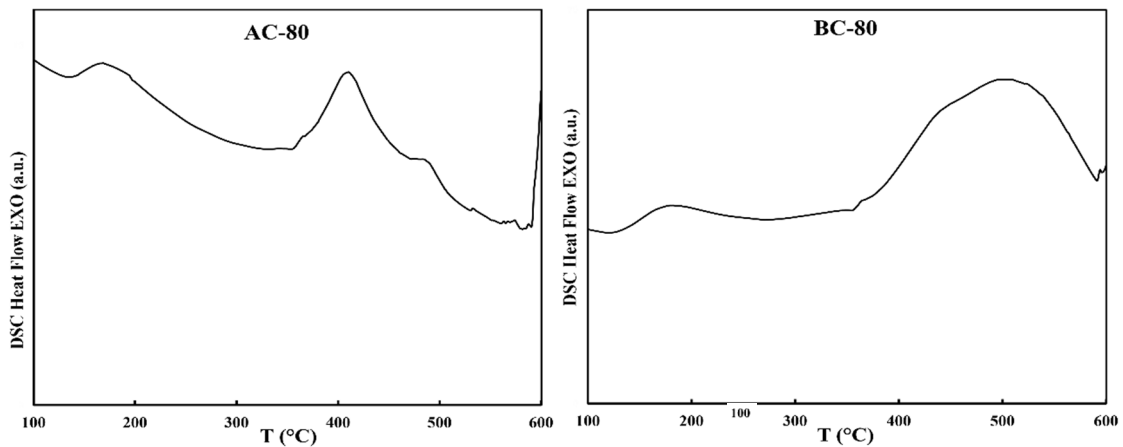


Figure 6. DSC curves of the compacted specimens AC-80 and BC-80.

Ultimately, exothermic processes corresponding to relaxation phenomena were found in samples compacted at temperatures below 200 °C. (AC-80 and BC-80). At 400 °C for AC-80 and 500 °C for BC-80, the exothermic peak related with the crystallization process was observed.

Likewise, we analyzed the kinetic of the mean crystallization process. The apparent activation for the main crystallization processes E , was determined using the Kissinger linear approach at heating rates of 5, 10, 20, 30, and 40 K. The Kissinger approach depends on the relation between peak temperature, T_p , and heating scan rate, β [23].

Figures 7 and 8 depict the linear fitting, while Table 4 provides the apparent activation energy. Using the following relation, the apparent activation energy is calculated from the slope of the linear fit where R is the universal gas constant:

$$E = -(\text{slope} \cdot R), \quad (1)$$

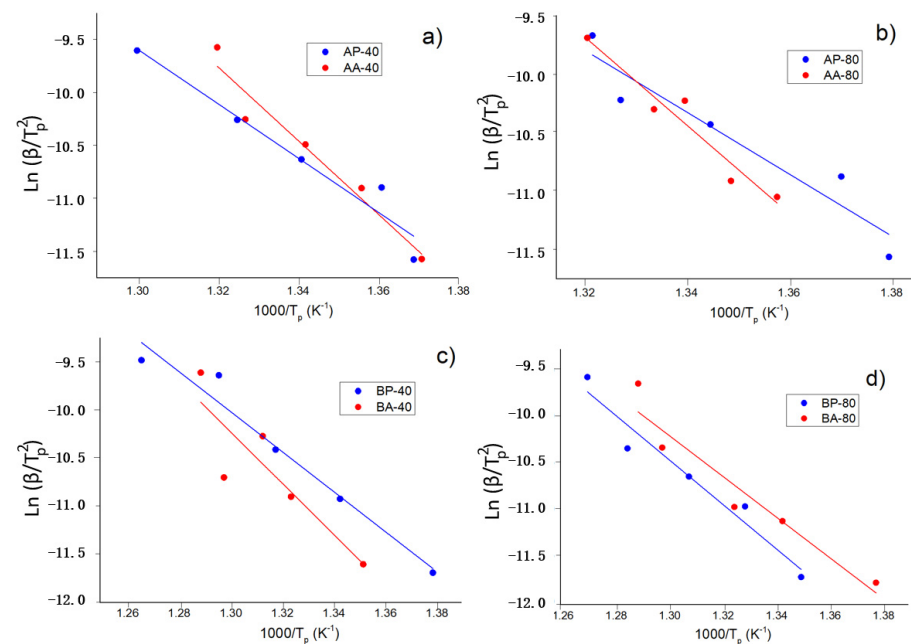


Figure 7. Kissinger linear fitting to determine the activation energy of the crystallization process of the powders and agglomerates (each experimental point corresponds to peak temperatures measured at 5, 10, 20, and 40 K/min). (a) Sample A-40h, (b) sample A-80h, (c) sample B-40h, and (d) sample B-80h. Activation energy (and its uncertainty) are given in Table 4.

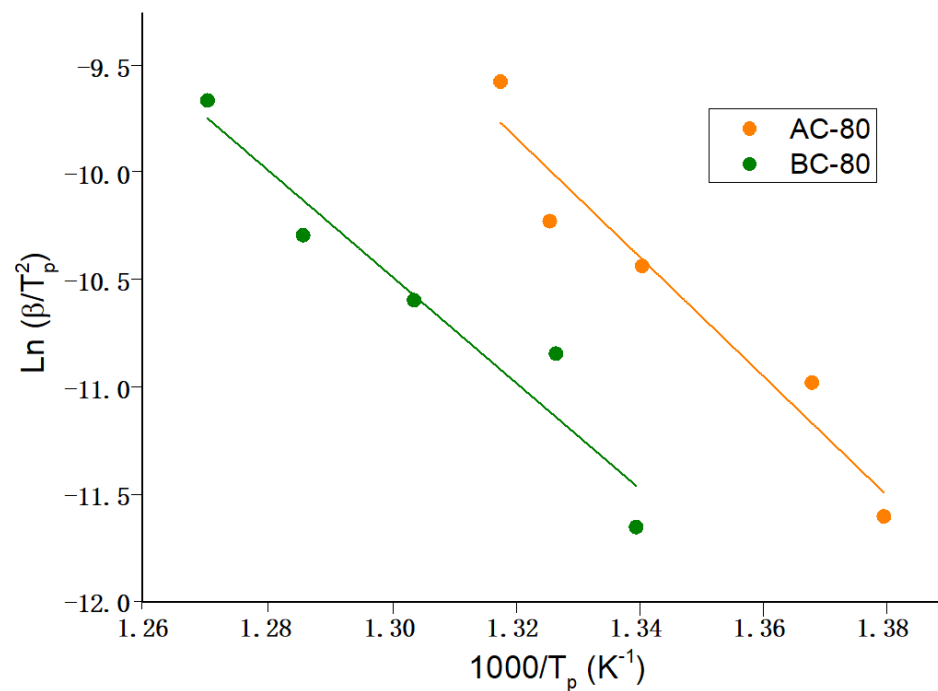


Figure 8. Kissinger linear fitting to determine the activation energy of the crystallization process of the compacted specimens (each experimental point corresponds to peak temperatures measured at 5, 10, 20, and 40 K/min). Activation energy (and its uncertainty) are given in Table 4.

Table 4. The activation energy of the main crystallization process in samples A and B.

Samples	Activation Energy/kJ mol ⁻¹	Samples	Activation Energy/kJ mol ⁻¹
AP-40	214 (6)	BP-40	173 (13)
AA-40	290 (4)	BA-40	222 (6)
AP-80	221 (18)	BP-80	199 (5)
AA-80	314 (9)	BA-80	182 (3)
AC-80	231 (20)	BC-80	205 (10)

For alloys A and B, the activation energies are higher in agglomerated particles than in powdered particles (at 40 h of milling). A possible explanation is diffusion and homogeneity inside the agglomerated particle stabilize against crystal growth, resulting in a higher activation energy. The same effect is found in alloy A milled for 80 h, whereas the opposite effect is found in alloy B milled for 80 h. According to the Rietveld refinement, the BA-40 and BP-40 powders have approximately the same amount of Nb(B), 3.8% and 3.0%, respectively. The alloy B milled for 80 h exhibited an opposite behavior. The activation energy of the agglomerated particles BA-80 is lower than that of the powders BP-80. One probable reason is the high amount of Nb(B) phase ~5% in the BA-80, which might have affected nucleation and/or crystal formation. The BP-80 presented approximately 1.9% of the Nb(B) phase. Probably, further annealing treatments (followed by XRD analysis) permit arriving at a complete explanation. Finally, this finding also demonstrates the B alloy samples have higher thermal stability than the A alloy samples. This behavior can be attributed to Nb having a greater heat of solution in the Fe matrix than Ni and Zr.

The activation energy of the compacted specimens is similar to the activation energy of the original powders as produced by mechanical alloying.

3.4. Magnetic Analysis

Likewise, we check the functional soft magnetic response. Magnetic hysteresis cycles at room temperature for alloys A and B are depicted in Figure 9. This type of hysteresis cycle is commonly observable for nanostructured materials with small magnetic domains.

This is due to the existence of structural defects inside grains as well as a high density of nanocrystals, which inhibits domain wall movements.

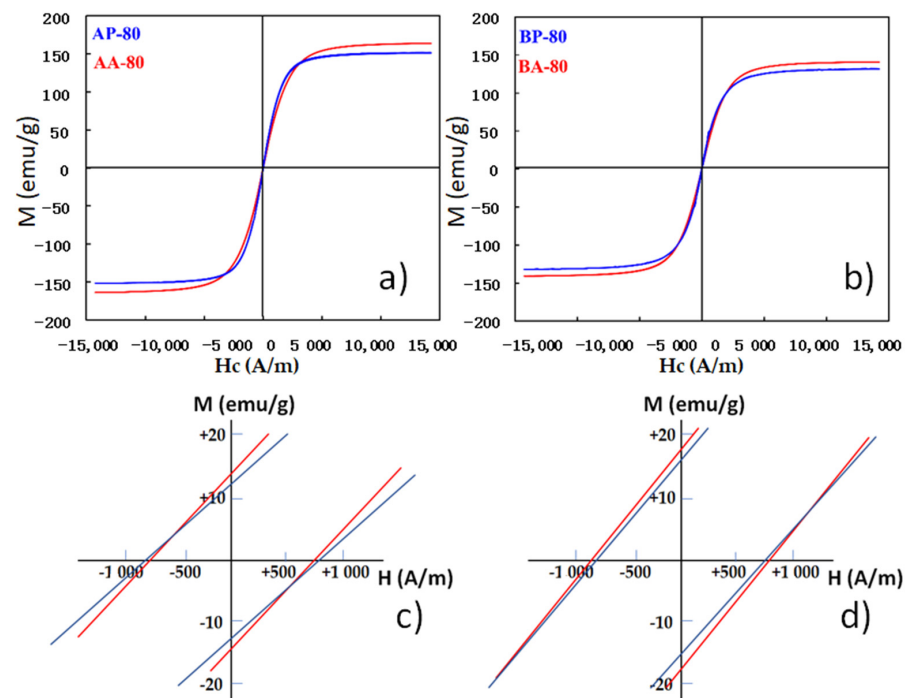


Figure 9. Magnetic hysteresis loops of samples: (a) AP-80, AA-80, (b) BP-80 and BA-80 and insets of the (0,0) region, (c). AP-80, AA-80, (d) BP-80 and BA-80.

Table 5 summarizes the results of the most relevant magnetic parameters: coercivity, remanent magnetization (M_r), and saturation magnetization (M_s). Magnetic characteristics are affected by sample microstructure parameters, such as crystallite size, particle shape, structural defects, etc. All the examined materials exhibit ferromagnetism in the nanocrystalline state at room temperature and have low coercivity (H_c) close to 770–833 A/m, which is one of the most significant criteria for a soft magnetic material; coercivity values lower than 1000 A/m are associated with soft magnetic materials [24]. The M_r/M_s values are comparable to those obtained in alloys produced by mechanical alloying [25].

Table 5. Relevant parameters determined from the magnetic hysteresis loops of samples AP-80, AA-80, BP-80, and BA-80.

Samples	H_c (A/m)	M_r (emu/g)	M_s (emu/g)	M_r/M_s
AP-80	770	13.8	151.0	0.091
AA-80	785	11.5	163.6	0.007
BP-80	812	18.1	131.3	0.138
BA-80	833	17.3	140.8	0.123

Regarding the bulk compacted specimens (after 80 h of milling), the coercivity was measured. The process consists of pressing at 600 MPa for 30 min in vacuum. The dimensions of the dies are 10 mm in diameter and about 3 mm thick. The analysis of the DSC scans suggests the best annealing temperature to favor the structural relaxation of the samples without inducing the crystal growth is close to 300 °C; annealing was performed at 300 and 600 °C. Higher temperatures are not recommended due to the formation of magnetically undesired intermetallic compounds, such as Fe_3B [26]. Table 6 shows the coercivity values of the associated average (five measurements in two samples of the same composition) and its statistical error.

Table 6. Coercivity, H_c , of compacted specimens at room temperature (RT) and after annealing (30 min) at 300 or 600 °C.

Sample	H_c (RT) /A m ⁻¹	H_c (300 °C) /A m ⁻¹	H_c (600 °C) /A m ⁻¹
Fe ₈₀ (NiZr) ₈ B ₁₂	827	784	1873
Fe ₈₀ Nb ₈ B ₁₂	845	807	2012

The coercivity values of compacted powder milled for 80 h (AC-80) were slightly higher than those of both specimens as milled. We found annealing for 1 h at 300 °C enhanced the coercivity soft behavior (by reducing their values about 5%). It was an expected result because DSC scans show a broad exothermic process linked to structural relaxation at low temperatures. Thus, annealing favors the reduction of dislocation, vacancies, and other defects typical of mechanically alloyed specimens [27]. Likewise, by annealing at 600 °C, the coercivity increases and the compacted material have a soft-hard behavior. This effect is due to the crystallization at temperatures higher than 300 °C (detected as exothermic peaks in DSC scans). The tendency of the results as a function of annealing temperature is the same to those found in specimens with similar composition in previous studies [28].

4. Conclusions

In a planetary high-energy ball mill, two nanocrystalline Fe₈₀(NiZr)₈B₁₂ (A) and Fe₈₀Nb₈B₁₂ (B) alloys were produced. The primary aim of the research was to see whether the powders and agglomerated particles generated during milling had the same microstructure, structural, thermal, and magnetic characteristics.

The structural analysis of the two alloy samples reveals the main phase of all samples was a BCC α -Fe solid solution. A low percentage of Nb(B) phase was identified in the B samples (powders and agglomerated particles). The A alloy's crystalline size varies between 11 and 14 nm, whereas for B alloys, it ranges between 8 and 12 nm. Agglomerated samples of both alloys had greater microstrain and dislocation density than milled powders.

Additional thermal analysis allowed us to identify the analyzed crystallization process is directly attributable to crystal growth, and thermal stability is higher in agglomerated samples except for the BP-80 that reveals higher stability than the BA-80, which we explained by the higher Nb(B) amount in BA-80.

Magnetic studies at room temperature of all samples milled after 80h revealed soft magnetic behavior with reduced coercivity values in the 770-833 A/m range. Annealing at 300 °C reduces the coercivity of compacted specimens.

Author Contributions: Conceptualization, J.-J.S.; formal analysis, K.Z., J.D., and W.B.M.; investigation, K.Z. and W.B.M.; writing—original draft preparation, K.Z. and J.-J.S. All authors have read and agreed to the published version of the manuscript.

Funding: This research was funded by Spanish Mineco under project PID2020-115215RB-C22.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Data can be requested to the authors.

Conflicts of Interest: The authors declare no conflict of interest.

References

1. Ma, H.J.; Wei, W.Q.; Bao, W.K.; Shen, X.B.; Wang, C.C.; Wang, W.M. Research progress and application prospect of Fe-based nanocrystalline soft magnetic alloys. *Rare Metal Mater. Eng.* **2020**, *49*, 2904–2912.
2. Sun, Y.; Li, J.W.; He, A.A.; Xie, L.; Dong, Y.Q.; Liu, Y.X.; Zhang, R.H.; Zhang, K.W. Influence of microstructure and anisotropy on the high-frequency soft magnetic properties of nanocrystalline FeSiBNbCuP alloys. *J. Magn. Magn. Mater.* **2022**, *560*, 169639. [[CrossRef](#)]

3. Suzuki, K.; Parsons, R.; Zang, B.; Onodera, K.; Kishimoto, H.; Shoji, T.; Kato, A. Nanocrystalline soft magnetic materials from binary alloy precursors with high saturation magnetization. *AIP Adv.* **2019**, *9*, 035311. [[CrossRef](#)]
4. Kolano-Burian, A.; Kowalczyk, M.; Grabias, A.; Radon, A.; Blyskun, P.; Warski, T.; Karpinski, M.; Hawelek, L.; Kulik, T. The influence of ultra-rapid annealing on nanocrystallization and magnetic properties of Fe_{76-x}Ni₁₀B₁₄. *J. Alloys Compd.* **2022**, *921*, 165943. [[CrossRef](#)]
5. Suryanarayana, C.; Al-Joubori, A.A.; Wang, Z. Nanostructured materials and nanocomposites by mechanical alloying: An overview. *Met. Mater. Inter.* **2022**, *28*, 41–53. [[CrossRef](#)]
6. Blázquez, J.S.; Ipus, J.J.; Conde, C.F.; Conde, A. Evolution of Fe environments in mechanically alloyed Fe-Nb-(B) compositions. *J. Alloys Compd.* **2014**, *615*, s555–s558. [[CrossRef](#)]
7. Suñol, J.J.; González, A.; Saurina, J.; Escoda, L.; Bruna, P. Thermal and structural characterization of Fe-Nb-B alloys prepared by mechanical alloying. *Mater. Sci. Eng. A* **2002**, *375–377*, 874–880. [[CrossRef](#)]
8. Alleg, S.; Hamouda, A.; Bensalem, R.; Suñol, J.J.; Greneche, J.M.; Azzaza, S. Solid state amorphization transformation in the mechanically alloyed Fe_{27.9}Nb_{2.2}B_{69.9} powders. *Mater. Chem. Phys.* **2010**, *122*, 35–40. [[CrossRef](#)]
9. Ipus, J.J.; Blázquez, J.S.; Franco, V.; Conde, A. Mechanical alloying of Fe_{100-x-y}Nb_xB_y (x = 5, 10; y = 10, 15): From pure powder mixture to amorphous phase. *Intermetallics* **2008**, *16*, 1073–1082. [[CrossRef](#)]
10. Alleg, S.; Ibrir, M.; Fenineche, N.E.; Azzaza, S.; Bensalem, R.; Suñol, J.J. Magnetic and structural characterization of the mechanically alloyed Fe₇₅Si₁₅B₁₀ powders. *J. Alloys Compd.* **2010**, *494*, 109–111. [[CrossRef](#)]
11. Chabi, T.; Alleg, S.; Suñol, J.J.; Bensebaa, N.; Azzaza, S.; Hlil, E.K. Effect of the boron content on the amorphization process and magnetic properties of the mechanically alloyed Fe_{92-x}Nb₈B_x powders. *J. Supercond. Nov. Magn.* **2019**, *32*, 893–901. [[CrossRef](#)]
12. Mnasri, C.; Cheminghi, M.; Nivot, C.; Tricoteaux, A.; Lorgouilloux, Y.; Dammak, M. Phase transformation and microstructural evolution in nanocrystalline Fe₇₁B₂₃Nb₆ alloy prepared by mechanical alloying. *Inter. J. Mater. Res.* **2020**, *111*, 307–315.
13. Ipus, J.J.; Blázquez, J.S.; Franco, V.; Lozano-Pérez, S.; Conde, A. Role of starting phase of boron on the mechanical alloying of FeNbB composition. *J. Alloys Compd.* **2013**, *553*, 119–124. [[CrossRef](#)]
14. Lamoglia, M.S.; Goncalves, P.H.; Pontes, A.M.P.; Serrano, L.B.; Silva, G.; da Silva, A.A.A.P. Effect of Process control agents on Fe-15at.%Nb powder during mechanical alloying. *Mater. Res. Ibero-Amer. J. Mater.* **2022**, *25*, e20210318. [[CrossRef](#)]
15. Hong, S.J.; Kao, P.W. Mechanical properties of AlSiC composites made by resistance sintering of mechanically alloyed powders. *Mater. Sci. Eng. A* **1991**, *148*, 189–195. [[CrossRef](#)]
16. Suryanarayana, C. Mechanical alloying: A critical review. *Mater. Res. Lett.* **2022**, *10*, 619–647. [[CrossRef](#)]
17. Yapici, G.G. simultaneous improvement in strength and ductility of severely deformed niobium alloy. *Mater. Lett.* **2020**, *279*, 128443. [[CrossRef](#)]
18. Izumi, K.; Seriya, C.; Okada, K.; Kudou, K.; Shishido, T. Mechanochemically alloyed assisted preparation of NbB₂ powder. *J. Eur. Ceram. Soc.* **2006**, *26*, 635–638. [[CrossRef](#)]
19. Malayathodi, R.; Sreekanth, M.S.; Deepak, A.; Dev, K.; Surendranathan, A.O. Effect of milling time on production of aluminium nanoparticle by high energy ball milling. *Int. J. Mech. Eng. Technol.* **2018**, *46*, 646–652.
20. Suryanarayana, C. Mechanical alloying and milling. *Prog. Mater. Sci.* **2001**, *46*, 1–184. [[CrossRef](#)]
21. Cotai, A.; Miraglia, S.; Neamtu, B.V.; Marinca, T.F.; Chicinas, H.F.; Isnard, O.; Chicinas, I. A comparative study of nanocrystalline Fe_{38.5}Co_{38.5}Nb₇P₁₅Cu₁ alloys obtained by mechanical alloying and rapid quenching. *Arch. Metall. Mater.* **2022**, *67*, 555–561.
22. Ipus, J.J.; Blázquez, J.S.; Franco, V.; Conde, A. The use of amorphous boron powder enhances mechanical alloying in soft magnetic Fe-Nb-B alloy: A magnetic study. *J. Appl. Phys.* **2013**, *113*, 17A330. [[CrossRef](#)]
23. Kissinger, H.E. Reaction kinetics in differential thermal analysis. *Anal. Chem.* **1957**, *29*, 1702–1706. [[CrossRef](#)]
24. Silveira, J.M.; Ferrara, E.; Huber, D.L.; Monson, T.C. Soft Magnetic materials for a sustainable and electrified world. *Science* **2018**, *362*, eaao0195. [[CrossRef](#)] [[PubMed](#)]
25. Triki, M.; Mechri, T.; Azzaz, H.; Azzaz, M. Characterization of nanostructured magnetic alloy based on Ni-Co-Mn produced by mechanical alloying. *J. Magn. Magn. Mater.* **2002**, *541*, 168514. [[CrossRef](#)]
26. Sun, Y.M.; Yu, W.Q.; Hua, Z. Effect of the thermal treatment on the microstructure and Magnetic property of Fe₆₈Zr₂₀B₁₂ alloy prepared by mechanical alloying. *Acta Phys. Pol. A* **2011**, *119*, 374–377. [[CrossRef](#)]
27. Ulianov, A.L.; Elskov, E.P.; Eremina, M.A.; Zagaimov, A.V.; Chulkina, A.A. Structural and phase transformations during heat treatment of the Fe_{71.4}Si_{14.3}C_{14.3} amorphous alloy prepared by mechanical alloying. *Phys. Met. Metallogr.* **2010**, *110*, 542–552. [[CrossRef](#)]
28. Carrillo, A.; Daza, J.; Saurina, J.; Escoda, L.; Suñol, J.J. Structural, thermal and magnetic analysis of Fe₇₅Co₁₀Nb₆B₉ and Fe₆₅Co₂₀Nb₆B₉ nanostructured alloys. *Materials* **2021**, *14*, 4542. [[CrossRef](#)]

Disclaimer/Publisher’s Note: The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.